- 1 Growth and biofilm formation of *Cupriavidus metallidurans* CH34 on
- 2 different metallic and polymeric materials used in spaceflight
- 3 applications
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Growth and biofilm formation of *Cupriavidus metallidurans* CH34 on different metallic and polymeric materials used in spaceflight applications

- 17 Bacteria biofilm formation and its complications are of special concern in 18 isolated structures, such as offshore stations, manned submarines and space 19 habitats, as maintenance and technical support are poorly accessible due to costs 20 and/or logistical challenges. In addition, considering that future exploration 21 missions are planned to adventure farther and longer in space, unlocking biofilm 22 formation mechanisms and developing new antifouling solutions are key goals in 23 order to ensure spacecraft's efficiency, crew's safety and mission success. In this 24 work, we explored the interactions between Cupriavidus metallidurans, a 25 prevalently identified contaminant onboard the International Space Station, and 26 aerospace grade materials such as the titanium alloy TiAl6V4, the stainless steel 27 AISI 316 (SS316) and Polytetrafluoroethylene (PTFE) or Teflon. Borosilicate 28 glass was used as a control and all surfaces were investigated at two different pH 29 values (5.0 and 7.0). Biofilms were almost absent on stainless steel and the 30 titanium alloy contrary to Teflon and glass that were covered by an extensive 31 biofilm formed via monolayers of scattered matrix-free cells and complex 32 multilayered clusters or communities. Filamentous extracellular DNA structures 33 were observed specifically in the complex multilayered clusters adherent to 34 Teflon, indicating that the employed attachment machinery might depend on the 35 physicochemical characteristics of the surface.
- 36 Keywords: Cupriavidus metallidurans; biofilm; eDNA; PTFE; stainless steel;
 37 titanium alloy

38 Introduction

39 Bacteria have the ability to attach to a surface and/or to each other and to form 40 multicellular communities embedded in a polymeric matrix known as biofilms 41 (Stoodley et al., 2002). This phenomenon is linked to an increased resistance to 42 antibacterial agents and despite decontamination processes, adherent bacteria are 43 problematic for many industries such as health care, drugs manufacturing, food

44 processing, oil extraction and electronics (Harrison et al., 2007; Mah and O'Toole, 45 2001; Van Houdt and Michiels, 2010). Biofilm formation and its complications are of 46 special concern in isolated structures, such as offshore stations, manned submarines and 47 space habitats, as maintenance and technical support are poorly accessible due to costs 48 and/or logistical challenges. Biofilms can cause surface corrosion, hardware 49 degradation, clots and failures in critical equipment such as piping or filters of life 50 supporting systems, water-recycling units, radiators and in extravehicular mobility units 51 (EMU). In fact, related to spaceflight applications, such incidents have already been 52 observed in the Salyut, Mir and Skylab stations, and continue to be a challenging threat 53 on board the International Space Station (ISS) (Diaz et al., 2019; Flemming, 1998; 54 Klintworth et al., 1999).

55 The main concern for biofilm control strategies in spacecraft revolves around 56 water storage and recycling systems as the initiation of biofilm formation depends 57 largely on humidity (water droplets, vapor, condensate, water/liquid pipes, liquid tanks). 58 For instance, numerous hazardous points have been identified in spacecraft wet systems 59 as potential entries for microbial contamination or to be prone to biofilm formation, and 60 most of them have been traced to the sanitary space, the Environmental Control and 61 Life Support System (ECLSS) of the spacecraft or of its EMU and in thermal control 62 systems (Roman and Minton-Summers, 1998; Schultz et al., 1991; Squire et al., 2014). 63 In general, the ECLSS is a complex network of tanks, pumps, pipes, hoses and filters 64 that allows the regeneration of potable water from the recovery of urine and humidity 65 condensate in an enclosed environment. On the ISS, it is especially complicated and it 66 extends on 29 orbital replacement units (ORU) where countless components made from 67 various materials (e.g. Stainless Steel 302, 303, 304, 316, 318-8, 6061 Aluminum, 68 TiAl6V4, Inconel 718) can be directly or indirectly interacting. Other materials of

69 spacecraft or space suits such as Teflon, Nafion, Fluorel, nylon, Nomex, Gore-tex,

nickel and boron-nickel alloys, 60NiTi, graphite, silver, polyurethane coated aluminum
and gold-anodized aluminum can also be in contact with humid environments (Squire et
al., 2014).

73 Both urine and humidity condensate collected in spacecraft can contain a large 74 number of molecules and various carbon sources that are suitable for microbial 75 proliferation. Even after processing, water samples from the Mir station had 0.005 to 76 0.023 g L-1 of total organic carbon. Analysis of humidity condensate in Shuttle 77 missions STS-45 and STS-47 showed that the total organic carbon levels ranged from 78 0.12 g L-1 to 0.7 g L-1 (National Research Council, 2000). These concentrations are 79 sufficient for microbial proliferation, especially for oligotrophic bacteria that can 80 survive at levels of carbons as low as 0.001 g L-1 (Ishida and Kadota, 1981). Next to 81 water and available nutrients, the characteristic surface properties of the materials, such 82 as hydrophobicity, surface energy and electron donor/acceptor potential, will play a 83 crucial role in orienting microbial adhesion.

84 Squire et al. (2014) indicate that due to technical limitations, the routine antimicrobial procedure (gamma irradiation or extended heat treatment at 87.7°C) 85 86 cannot be applied to all ORUs elements before launch. In fact, 5 from 16 items that are 87 launched wet or containing water are not subjected to disinfection. This could create 88 favorable conditions for inflight microbial growth and potential biofilm formation that 89 could spread inside the wet system. Also microbial monitoring campaigns onboard the 90 ISS showed recurrent microbial contamination events (Van Houdt and Leys, 2020; Zea 91 et al., 2020). Although biofilms in water systems are interacting multispecies 92 communities (Thompson et al., 2020; Yang et al., 2021), one species that attracted 93 attention is Cupriavidus metallidurans. A Gram-negative bacterium belonging to the

94	Burkholderiaceae family that has been detected from 2009 to 2019 in almost all
95	samples coming from the wastewater tank, the potable waterbus or the condensate
96	(Mijnendonckx et al., 2013; Zea et al., 2020). This facultative chemolithotrophic motile
97	microbe shows resistance to a broad range of metals, including silver used as
98	disinfectant onboard ISS, and is able to adapt to various harsh conditions, including low
99	nutrients environments (Maertens et al., 2020; Mijnendonckx et al., 2019;
100	Mijnendonckx et al., 2013; Van Houdt et al., 2021; Zhang et al., 2018). In addition,
101	bacteria are exposed to specific conditions (e.g. microgravity and cosmic radiation)
102	during spaceflight (Acres et al., 2021; Bijlani et al., 2021; Horneck et al., 2010; Huang
103	et al., 2018), which have also been studied for C. metallidurans type strain CH34
104	(Byloos et al., 2017; Leys et al., 2009) (De Gelder et al., 2009; Leroy et al., 2010).
105	Furthermore, it is used to explore future spaceflight applications such as testing
106	antimicrobial surfaces (Siems et al., 2022) as well as biomining and bioremediation
107	(Byloos et al., 2017; Cockell et al., 2020; Santomartino et al., 2020). It is therefore a
108	representative of the contaminant species found in humid spacecraft systems as well as
109	a microbe with potential extra-terrestrial applications.
110	As future exploration missions are planned to adventure farther and longer in
111	space, unlocking biofilm formation mechanisms and developing new antifouling
112	solutions suitable for use in such a challenging environment are goals of key importance
113	in order to ensure spacecraft's efficiency, crew's safety and mission success. In this
114	work, we aim to explore the interactions between C. metallidurans and aerospace grade
115	materials such as titanium alloy TiAl6V4, stainless steel AISI 316 (SS316) and
116	polytetrafluoroethylene (PTFE) or Teflon. The impact of these materials on planktonic

117 growth and biofilm formation, in addition to its biodeterioration effect are investigated

118 at different pH values.

119 Material and methods

120 Bacterial strains, media and culture conditions

- 121 C. metallidurans CH34 was routinely grown in Tris-buffered mineral medium
- 122 (MM284) (Mergeay et al., 1985) supplemented with 2 g L-1 sodium gluconate as the
- sole carbon source. The final pH was adjusted to 5.0 or 7.0 with HCl 37%. Although
- 124 gluconate can chelate metal ions (Gyurcsik and Nagy, 2000) and could impact biofilm
- 125 biomass and production of extracellular polymeric substances (Liu et al., 2021; Luo et
- 126 al., 2019; Pal and Paul, 2013), it is readily consumed by C. metallidurans and as such
- 127 would have a limited impact.

128 **Preparation and setup of biofilm experiments**

- 129 Materials tested included coupons of borosilicate glass (75 mm x25 mm x1 mm,
- 130 stainless steel AISI 316 (used in ECLSS tanks and pipes, pumps and separators (Squire
- 131 et al., 2014); 80 mm x 5 mm x 0.5mm), Teflon (used in hoses of thermal control
- 132 systems, valves, bladders, insulation and gas separators (Squire et al., 2014; Wieland
- and Center, 1998); 18 mm x 13 mm x 3 mm) and titanium alloy TiAl6V4 (used in the
- 134 outer shell of water tanks and tubing (Petala et al., 2020); 25mm x 25 mm x1 mm) were
- 135 cleaned with deionized water and sonicated in 70% ethanol for 15 min. Next, coupons
- 136 were rinsed with deionized water and autoclaved. Precultures of C. metallidurans CH34
- 137 were grown at 30°C on an orbital shaker (150 rpm) until an optical density (OD600) of
- 138 0.2 in MM284 (pH 7.0). Each sterilized coupon was then placed in a 50 mL conical
- 139 centrifuge tube to which 15 mL of MM284 and 350 μ L preculture were added. Tubes
- 140 were then placed on a tilting (15°) shaker and incubated at room temperature ($23 \pm 2^{\circ}$ C)
- 141 and 1 rpm (turbulent flow conditions) for 168 hours.

142 Analysis of growth and biofilm formation

143 Planktonic growth was evaluated by measuring OD600 every 24 h. Biofilm formation

- 144 was assessed using the crystal violet (CV) dye that binds to DNA and proteins. A
- solution of 0.1% CV was prepared (0.1 g of CV in 2 mL 95% ethanol and 98 mL of
- 146 deionized water). At the end of the experiment, coupons were gently removed and softly
- 147 rinsed 3 times with deionized water. Coupons were then air-dried and covered with the
- 148 CV solution for 15 min. Afterwards, the CV solution was discarded and coupons were
- 149 rinsed 3 times with deionized water. Coupons were left to air dry before quantifying the
- amount of CV by solubilizing in 4 mL of 95% ethanol and measuring absorbance at 620
- 151 nm. Measurements were normalized by subtracting the average of abiotic controls.
- 152 For biofilm viable counts, coupons were rinsed twice with sterile saline solution (0.85%
- 153 NaCl). Next, coupons were placed in a new sterile tube with 15ml of saline solution and
- 154 sonicated (30s, 35kHz) in a sonication bath (Elma, Germany) and subsequently
- 155 vortexed (20 min, 2700 rpm). Cell suspensions (100 μL) of a serial ten-fold dilution in
- 156 saline were spread on MM284 agar plates and incubated at 30°C. Colony forming units
- 157 (CFU) were counted after 3 days.

158 Scanning Electron Microscopy (SEM)

159 Biofilms were examined under the SEM at the end of the experiment (168 h). Surfaces

160 were gently rinsed with deionized water to remove non-adherent cells. Samples were

161 then fixed for 2 hours in Bouin's solution (Jacobsen et al., 1980). Next, samples were

162 dehydrated by immersion in an ethanol graded series as follows: 30 min in 70% ethanol,

- 163 overnight in 70% ethanol, 30 min in 70% ethanol, 2 times 30 min in 90% ethanol, and 1
- 164 h in 100% ethanol. After dehydration, samples were dried with liquid CO2 in a critical
- 165 point dryer. Finally, they were gold-coated in a sputter coater and placed in a JSM-
- 166 7200F microscope (JEOL, Japan).

167 **DNase treatment**

168 Biofilms were grown as previously described (only at pH 7 to avoid interference with

- 169 DNase activity). Growth medium was discarded and coupons were gently rinsed with
- 170 sterile deionized water to remove loosely attached cells. Rinsed coupons were immersed
- 171 in 3 ml of 5 μg mL-1 DNaseI (Qiagen, Germany) and incubated 1 h statically at 37°C.
- 172 Next, samples were rinsed, stained with CV, dried at room temperature and observed
- 173 with an automated inverted fluorescence microscope (TE2000-E; Nikon, Tokyo, Japan)
- equipped with a Cy5 filter set (excitation passband: 590-650 nm; emission: 660-
- 175 740nm).

176 Inductively Coupled Plasma - Atomic Emission Spectroscopy (ICP-AES)

- 177 The concentration of iron and aluminum leached from SS316 and TiAl6V4,
- 178 respectively, were determined in the culture medium and the planktonic cells with ICP-
- 179 AES. At the end of the experiment (168 h), coupons were removed from the culture
- 180 medium and bacteria in suspension were pelleted via centrifugation at 10.000 g for 2
- 181 min. One mL of the supernatant was sampled and diluted with 14 mL of 5% HNO3.
- 182 The pellets were resuspended in 1 mL of filtered deionized water and 2 mL of
- 183 concentrated HNO3 was added to each sample and left to digest overnight at room
- 184 temperature. Finally, 12 mL of 5% HNO3 was added. For abiotic controls, samples
- 185 were centrifuged and 1 mL supernatant was diluted in 14 mL of 5% HNO3. All samples
- 186 were stored at $+5^{\circ}$ C until analysis.

187 *Contact-angle measurements*

- 188 Coupons were incubated 3 h in sterile MM284 (final pH of 5 or 7). Next, coupons were
- removed from the medium and left to air-dry before analysis. Contact angle
- 190 measurement was performed using a VCA optima goniometer (AST Products, Billerica,

191 USA) according to the static drop method. Three μ L of deionized water were deposited 192 on the dry substratum and six measurements were made for each sample. Images were 193 analyzed with VCA Optima software (AST Products, Billerica, USA). The surface was 194 considered hydrophilic or hydrophobic when the contact angle was less or more than 195 90°, respectively.

196 Statistical analysis

197 Statistical comparison of two samples was performed using the Student's t-test.

198 Statistical comparison of multiple samples was performed using a one-way ANOVA

analysis, followed by a post-hoc Tukey test. A p-value less than 0.05 was consider

200 statistically significant.

201 **Results and Discussion**

202 Surface characterization

203 The surfaces of the uninoculated clean materials were first analyzed under the SEM. 204 SS316 appeared regular and streaked with parallel thin bands (Figure 1b). Teflon 205 appeared mainly smooth and crossed occasionally by stripes or stretches of nearly 5-µm 206 width (Figure 1c). The glass surface was very smooth and free from any special 207 topographic features (Figure 1a), whereas TiAl6V4 was rough and covered with crests 208 and pits of 10 to 30 µm wide (Figure 1d). Contact angle measurements showed that 209 glass was highly hydrophilic at pH 5.0 and 7.0, and no significant differences in contact 210 angles were observed (Figure 2). Glass was the most hydrophilic substratum with very 211 low contact angles that did not exceed $25.31 \pm 6.45^{\circ}$. Teflon was the most hydrophobic 212 material with contact angles not less than $109.11 \pm 6.61^{\circ}$ (Figure 2) and SS316 was 213 amphiphilic. Finally, although contact angles of TiAl6V4 varied significantly between 214 pH 5.0 and pH 7.0, it remained hydrophilic (Figure 2). Since the growth medium

(MM284) contained various ions and a carbohydrate (gluconate), contact angles were
also measured in pure water. Indeed, MM284 components significantly decreased the
contact angles for glass and TiAl6V4 at pH 5 and pH 7, and for SS316 at pH 5 (Figure
2). At the contrary, a significant increase in contact angle was observed for Teflon
between the conditioning in MM284 pH 7.0 and in pure water.

Planktonic growth of C. metallidurans CH34 in the presence of the test surfaces

222 It is clear that bacteria show variable sensitivities to pH. Whereas neutrophilic bacteria 223 prosper in circumneutral pH, acidophilic and alkaliphilic ones prefer opposite sides of 224 the pH scale. In addition, the presence of macromolecules, nutrients and sensitizing 225 compounds can change the bacterial response to the surrounding pH. Non-optimal pH 226 or pH shifts are stressful for bacteria, it can unbalance the cellular homeostasis, disrupt 227 membrane integrity and fluidity and inhibit key reactions involved in bacterial growth 228 or survival (Guan and Liu, 2020; Padan et al., 2005). In addition, pH also plays a critical 229 role in the readjustment of material surface properties by influencing hydrophobicity 230 and electrostatic forces. It can slow or accelerate the corrosion of some materials and can enhance the leaching of their composing elements. Therefore, we performed the 231 232 experiments at neutral (pH 7.0) and acidic (pH 5.0) conditions. Without the presence of 233 the test surfaces C. metallidurans showed comparable growth at pH 5.0 and 7.0 (Figure 234 3). In addition, none of the test surfaces, and any leached elements, affected planktonic 235 growth at pH 5.0 or 7.0 (Figure 3). Leaching of chromium and nickel ions, and 236 aluminum and vanadium ions in biological media have been reported for stainless steel 237 and titanium alloy, respectively (Berggren et al., 2004; Hedberg and Odnevall 238 Wallinder, 2015; Herting et al., 2006; Mumme et al., 2005; Zhou et al., 2011).

239 Adhesion and biofilm formation of C. metallidurans CH34 on the test surfaces 240 Although prolonged contact with the test materials did not affect planktonic growth, the 241 physicochemical differences between the test surfaces did significantly influence 242 adhesion of C. metallidurans CH34. Crystal violet staining showed that Teflon and 243 glass were more prone to C. metallidurans CH34 adhesion and biofilm formation than 244 TiAl6V4 and SS316 (Figure 4, Supplementary Figure 1) with Teflon having 245 significantly more adherent biomass than glass both in pH 5 and pH 7.0. Determination 246 of viable biofilm cell numbers by plate count was also performed for glass and Teflon (pH 5), and corroborated the quantification by CV staining (Supplementary Figure 2). 247 248 Adhesion to glass also varied according to the pH as the CV-stained adherent biomass 249 on glass at pH 5.0 was nearly two-fold more important than at pH = 7.0. For Teflon, 250 adhesion was independent of the pH. A previously reported theoretical model predicted 251 that pH variations could influence S. aureus adhesion to glass because it alters its 252 hydrophilicity/hydrophobicity (Hamadi et al., 2009). In our experiment, the surface 253 properties of glass were not significantly affected during the experiment (Figure 2), 254 therefore, other factors are putatively responsible for the observed difference between 255 pH 5.0 and 7.0 (Sheng et al., 2008). It is generally assumed that hydrophobic (reduces the strength of repulsion 256

257 forces between the substratum and bacteria) and rough (creates areas of low shear 258 stress) surfaces are more attractive for microbial adhesion than hydrophilic and smooth 259 ones (Zheng et al., 2021). Indeed, the hydrophobic Teflon surface showed the most 260 adhesion compared to the other materials. High adhesion on Teflon has previously been 261 reported for Pseudomonas aeruginosa (Alfa et al., 2017), Salmonella spp. and Listeria 262 monocytogenes (Sinde and Carballo, 2000). Although glass is very hydrophilic, with contact angles from $12.3 \pm 2^{\circ}$ to $25.3 \pm 6.5^{\circ}$, it was the second most attractive material 263 264 for C. metallidurans CH34 biofilm formation. The capacity of bacteria to attach both to

265 hydrophilic and hydrophobic surfaces has already been observed for *Staphylococcus*

epidermidis and in *P. aeruginosa* (Cerca et al., 2005; Shelobolina et al., 2018), and was
corroborated here for *C. metallidurans*.

268 Stainless steel 316, which was significantly more hydrophobic than glass (Figure 269 1), showed limited C. metallidurans CH34 biofilm formation at pH 5.0 and no biofilm 270 formation at pH 7.0. (Figure 4). Interestingly, biofilm formation on stainless steel 304 in 271 mineral water has recently been documented (Maertens et al., 2020). Studies 272 demonstrated the superior capacity of SS316 over SS304 in repelling microbial 273 adhesion in potable water (Percival et al., 1997). These differences could be explained 274 by the different composition of SS316, which contains 2 to 3% of Molybdenum 275 (Percival, 1999), or by specific growth conditions. In addition, SS316 is known to be 276 less vulnerable to corrosion and pitting in the presence of chloride or low pH 277 environments, and also to microbiologically influenced corrosion (Wang et al., 2021). 278 The introduction of Mo in the alloy's formula has therefore significant consequences on 279 the physicochemical properties of the material. 280 Titanium alloy contact angles $41.6 \pm 5.7^{\circ}$ to $53.8 \pm 4.7^{\circ}$ were significantly 281 higher than glass. In addition, the alloy showed a rough surface rich in large pits that 282 could have been ideal sites for bacterial cells seeking protection from shear forces. 283 Nevertheless, almost no C. metallidurans CH34 biofilm formation was detected (low 284 quantification with large deviations). However, TiAl6V4 surfaces are not invulnerable 285 to bacterial adhesion and can be colonized by Serratia spp., sulphur-oxidizing and 286 sulphate-reducing bacteria (Cwalina et al., 2017) and by clinical isolates of 287 Streptococcus, Staphylococcus and Escherichia coli (Wang et al., 2018). Therefore, 288 properties such as surface hydrophobicity and roughness alone or even combined are 289 not the only factors that mediate bacterial adhesion and the development into a mature

biofilm. In fact, Gyo et al. (2008) coined that the relationship between hydrophobicity
and biofilm formation is controversial and put forward that the adhesion process in
immersed biofilms is even more complicated because of potential anomalies between
surface properties and submerged conditions.

294 Biofilm architecture

295 Biofilms were also observed under the SEM. On glass, C. metallidurans CH34 296 biofilms grew either as single cells, cells clustered in a polymeric extracellular matrix or 297 dense globular clusters (Figure 5a, b, c and d). On Teflon, the density of single adherent 298 cells was higher than for glass, which is in agreement with CV quantification. The cell 299 clusters were also larger than those observed on glass. In addition, Teflon-adherent 300 clusters were held together and anchored to the surface by a complex filamentous web 301 (Figure 5e, f, g and h). The web was composed of interconnected thin strands of 20 to 302 30 nm width that can reach nearly two µm in length (Figure 5g and h). Interestingly, 303 these filamentous networks were observed neither in glass-adherent C. metallidurans 304 CH34 biofilms (Figure 5a, b, c and d) nor in planktonic cells grown in the presence of 305 Teflon and of glass (Supplementary Figure 3). Such structures were reported in biofilms 306 of Enterococcus faecalis (Barnes et al., 2012), P. aeruginosa (Wang et al., 2015), 307 Ralstonia solanacearum (Minh Tran et al., 2016) and Streptococcus mutans (Kim et al., 308 2018), and are most likely composed of extracellular DNA (eDNA). The latter is a key 309 stabilizing element in many bacterial biofilms (Campoccia et al., 2021; Okshevsky and 310 Meyer, 2015; Panlilio and Rice, 2021). The release of eDNA is facilitated either 311 through cellular lysis or via membrane vesicles (Panlilio and Rice, 2021). Indeed, we 312 observed that biofilm cells were covered with outer membrane vesicles (OMVs) of 20 313 to 50 nm width that putatively could release proteins or nucleic acids (Figure 5h). The 314 resulting multilayered structure was capable of capturing solid components present in

315 the media such as suspended crystals or debris, reinforcing further the biofilm (Figure 316 5f). In addition, unique polar protrusions connecting cells in an organized manner were 317 also observed (Supplementary Figure 4). Overall, SEM observations revealed two 318 distinguishable adhesion patterns for C. metallidurans CH34, i.e. monolayers of 319 scattered matrix-free cells and complex multilayered clusters or communities. 320 As previously described, glass and Teflon differ drastically in their 321 physicochemical properties and it is possible that these dissimilarities induced a 322 different attachment strategy. In fact, it has been shown that the presence of eDNA 323 significantly increased the cell envelope hydrophobicity of S. epidermidis and allowed 324 for a stronger preference to hydrophobic surfaces through acid-base interactions (Das et 325 al., 2010). Likewise, for C. metallidurans CH34, eDNA putatively mediated attachment 326 to the hydrophobic surface of Teflon and could be disadvantageous for the adhesion to 327 hydrophilic glass. Many bacteria can actively regulate the release of eDNA (Ibanez de 328 Aldecoa et al., 2017; Minh Tran et al., 2016) and we hypothesize that this is also the 329 case for C. metallidurans CH34. To investigate the role of the observed net-like 330 frameworks further, we exposed mature C. metallidurans CH34 biofilms on Teflon to 331 DNaseI. This treatment apparently reduced the thickness and cell density at the edge of 332 the cluster, suggesting that eDNA has a role in aggregation and cell build-up at the 333 boundaries of the biofilms (Figure 6). Nevertheless, DNase I treatment did not affect the 334 total adherent biomass quantified via CV (data not shown) and appeared to have a 335 limited effect on the mature biofilms. In addition, it has been shown that the efficiency 336 of DNase treatment was minimal in mature biofilms of E. coli, Klebsiella pneumoniae, 337 P. aeruginosa and S. mutans, because mature biofilms accumulated more Z-form eDNA 338 that is, unlike the B-form, resistant to DNaseI (Buzzo et al., 2021).

339 Elemental release from the test surfaces

340 Next, the possible release of elements from SS316 and TiAl64, and the role of 341 C. metallidurans CH34 herein was investigated. SS316 is mainly composed of iron (at 342 least 65%), next to chromium (16-18%), nickel (10-14%) and molybdenum (2-3%). A 343 significant higher Fe release was observed in the media of abiotic controls at pH 7.0 344 compared to when SS316 was incubated with bacteria (Figure 7, bottom). Gluconate 345 probably acted as a chelator and released Fe from the surface (Sawyer, 1964). Fe 346 leaching from the surface in the presence of C. metallidurans is probably reduced as 347 gluconate is consumed. No significant differences in Fe content of the bacterial biomass 348 were observed in the presence or absence of SS316 (Figure 7, bottom). The TiAl6V4 349 alloy contains next to titanium, at least 6% aluminum and 4% vanadium. No significant 350 differences in the aluminum levels of the cell-free supernatant of the biotic samples and 351 abiotic controls were found at pH 5.0 and pH 7.0. Interestingly, when cells were 352 incubated with the TiAl6V4 surface at pH 7.0 the biomass contained nearly 2 times 353 more aluminum compared to the biotic control. This difference was not observed at pH 354 5.0. Overall, this indicated more release of Al from the titanium alloy at pH 7.0. Finally, 355 elemental release from Teflon, which is considered one of the most chemically inert 356 polymers and non-degradable by microorganisms under aerobic and anaerobic 357 conditions, was not investigated.

358 Conclusions

Our study on biofilm formation of *C. metallidurans* CH34 on four spacerelevant surfaces showed that stainless steel SS316 and the titanium alloy TiAl6V4 were less prone to biofouling in the tested conditions (i.e. mineral growth medium with gluconate as sole carbon source in turbulent flow conditions). The use of these materials for spaceflight applications, such as water management systems, can thus be beneficial 364 to prevent the build-up of Cupriavidus metalliduransbiofilms. However, as the 365 contamination in such systems is composed of multiple interacting species, multispecies 366 experiments should be scrutinized in a next step. Interestingly, in the water-dispensing 367 unit designed lately by NASA, stainless steel sections in contact with water were 368 replaced by tubing in Teflon. As we demonstrated here, Teflon surfaces are highly 369 attractive for *Cupriavidus metallidurans* and the frequent dispensing function of such 370 equipment represents a potential entry point for this resilient bacterium (Maryatt, 2018). 371 It is reasonable to assume that coating and grafting of the studied materials with 372 antimicrobials may improve their performance in time as well as versus other bacterial contaminants. 373

374 Our experiments also revealed that surface hydrophobicity and roughness alone 375 or even combined are not the only factors that drive bacterial adhesion. Despite the 376 differences in their physicochemical characteristics, both hydrophobic fluoropolymeric 377 (Teflon) and hydrophilic borosilicate (glass) material allowed extensive biofilm 378 formation via monolayers of scattered matrix-free cells and complex multilayered 379 clusters or communities. In addition, C. metallidurans CH34 likely employs a distinct 380 attachment machinery depending on the physicochemical characteristics of the surface. 381 We showed that filamentous structures described as extracellular DNA networks were 382 specific to the complex multilayered clusters growing on Teflon.

These biofilms were only locally sensitive to the enzymatic action of DNaseI, suggesting that the filamentous eDNA mesh may have a specific role in CH34 biofilm's architecture. As such, the use of enzyme-based antifouling products that induce the hydrolysis of DNA should be carefully studied (Okshevsky et al., 2015). These products are reported to weaken biofilm interactions and increase the permeability to antimicrobial treatments (Okshevsky et al., 2015). For developing anti-biofilm protocols it is important to take into account that bacterial adhesion is a complex phenomenon

390 where attached cells from multiple different species can be in different phenotypic and

391 metabolic states and consequently have different sensitivity to treatments.

Finally, it is important to remind that water recycling and recovery systems are composed of numerous other materials and all are interacting indirectly through the flowing liquid. Therefore, possible consequences of the observed iron leaching and aluminum accumulation should be investigated in long-term experiments.

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- 669 Figure 1. SEM images of the test materials, i.e. glass (a), stainless steel SS316 (b),
- 670 Teflon (c) and titanium alloy TiAl6V4 (d).
- Figure 2: Contact angle measurements of test materials in MM284 at pH 5.0 (red) and
- 672 pH 7.0 (blue), and water (green). The average values of three independent experiments
- 673 (n=6) with standard deviations are shown. Brackets indicate which samples within the
- 674 surface set are statistically different (p < 0.05) based on ANOVA and post-hoc Tukey.
- 675 Figure 3: Growth of *C. metallidurans* CH34 in the absence (black) or presence of glass
- 676 (red), stainless steel SS316 (blue), titanium alloy TiAl6V4 (green) or Teflon (purple)
- 677 surface. The average values of three independent experiments with standard deviations
- are shown.
- 679 Figure 4. Crystal violet quantification of C. metallidurans CH34 biofilm formation after
- 680 168 hours at pH 5.0 (red) and 7.0 (blue) on the test materials. The average values of
- three independent experiments with standard deviations are shown. For each pH value,
- 682 the different letters above the error bars indicate significant differences (p < 0.05) based
- on ANOVA and post-hoc Tukey.
- Figure 5. Representative SEM images of *C. metallidurans* CH34 biofilms on glass (a, b,
 c and d) and Teflon (e, f, g and h).
- 686 Figure 6. Effect of DNAseI treatment (b) versus control (a) on Teflon-grown
- 687 C. metallidurans biofilms visualized by crystal violet staining.
- 688 Figure 7. ICP-MS analysis of Al release from titanium alloy TiAl6V4 (top panel) and
- 689 Fe from stainless steel SS316 (bottom panel) by C. metallidurans CH34 (left) and
- 690 uninoculated growth medium (middle), and compared with their content in
- 691 C. metallidurans cells (right) for both the liquid (light grey) and bacterial cell (dark
- 692 grey) fraction. The average values of three independent experiments (n=3) with
- 693 standard deviations are shown.
- 694 Supplementary Figure 1: Crystal violet staining illustrating biofilm formation by *C*.
- 695 *metallidurans* CH34 on glass (left panel) and Teflon (right panel) after 168h in pH 5.0.
- 696 Supplementary Figure 2: Crystal violet quantification (dark grey) and plate count (light
- 697 grey) of C. metallidurans CH34 biofilm formation after 168 hours at pH 5.0 on glass

- and Teflon. The average values of three independent experiments with standard
- 699 deviations are shown. The different letters above the error bars indicate significant
- 700 differences (p < 0.05) based on ANOVA and post-hoc Tukey.
- 701 Supplementary Figure 2: (a) Planktonic cells of *C.metallidurans* CH34 cultured in the
- 702 presence of Teflon versus (b) planktonic cells cultured in the presence of glass.
- 703 Supplementary Figure 3: Polar protrusions and linear organisation of *C.metallidurans*
- 704 CH34 cells adherent to Teflon.